THE STEREOSPECIFIC PREPARATION OF METHYL FARNESOATE ${\rm AND~SYNTHETIC~PRECURSORS~Of~C}_{18} {\rm -~AND~C}_{17} {\rm ~JUVENILE~HORMONES}$

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The new method for the stereospecific preparation of 1,5-diene units was successfully applied to the syntheses of methyl farnesoate and the synthetic precursors of $\rm C^{}_{18}$ - and $\rm C^{}_{17}$ juvenile hormones, 15 and 20.

We have recently reported a stereospecific preparation of ethyl geranate from 3-methyl-2-butenyl bromide according to the following scheme. The key steps of this sequence are (1) the trans addition of benzenethiol to the α,β -acetylenic

ester, $^{2)}$ and (2) the stereospecific replacement of the phenylthio group by methyl group. $^{3)}$ This method, formally an overall trans addition of methylmagnesium iodide to the α , β -acetylenic ester, is expected to provide a general route to the stereospecific synthesis of trans 1,5-diene units which represent structural moiety found in many naturally occurring products. In this communication we wish to describe the stereospecific preparation of synthetic precursors of C_{18} - and C_{17} juvenile hormones, $\underline{15}$ and $\underline{20}$, and methyl farnesoate by the repetitive application of the above sequence.

When methyl 2-butynoate was allowed to react with sodium benzenethiolate in methanol-water (4 : 1), the trans addition took place predominantly and methyl (2)-3-phenylthio-2-butenoate, $\underline{1}$, (bp 128 \sim 130°C/3mmHg), was isolated in 77%

yield. The stereospecific conversion of $\underline{1}$ to the ester $\underline{2}$ was carried in 73% yield by the coupled use of ethylmagnesium bromide and cuprous iodide in tetrahydrofuran

$$CO_2Me \longrightarrow CO_2Me \longrightarrow CO_2Me \longrightarrow OH$$

at -78°C. Aluminum hydride reduction of $\underline{2}$ in ether afforded the allylic alcohol $\underline{3}$, bp 79 \sim 81°C/40mmHg, in 65% yield (purity > 99%). By the separate route, the same alcohol $\underline{3}$ was prepared stereospecifically by the following sequence consisting of (1) thioacetalization of ethyl 3-oxopentanoate, (2) base-catalyzed elimination of

benzenethiol from the thioacetal $\underline{4}$, (3) methylation of $\underline{5}$ with methylmagnesium bromide and cuprous iodide, and (4) aluminum hydride reduction of the ester $\underline{6}$.

The homologation of the C_6 -alcohol $\underline{3}$ to the C_{12} -alcohol $\underline{12}$ was achieved by the same procedure described in the preparation of ethyl geranate. The alcohol $\underline{3}$ was converted to the bromide $\underline{7}$, which was in turn treated with propargylmagnesium bromide in ether at 0°C to give the terminal acetylene $\underline{8}$ accompanying a small amount of the allenic isomer. A tetrahydrofuran solution of $\underline{8}$ was injected with calculated amount of n-butyllithium at 0°C, followed by the addition of methyl

chloroformate at -78°C. After the reaction mixture was stirred at that temperature for 1 hr and at room temperature for 3 hr, the acetylenic ester $\underline{9}$ was obtained in 50% yield based on the alcohol $\underline{3}$. The vinyl sulfide $\underline{10}$, $\underline{4}$) produced in 78% yield from the reaction of $\underline{9}$ with benzenethiol under basic condition, was alkylated

stereospecifically by ethylmagnesium bromide and cuprous iodide to form the dienic ester $\underline{11}$, bp 80 \sim 81°C/0.4mmHg, in 90% yield (purity 97%), and further converted to the C_{12} -alcohol $\underline{12}$ quantitatively (purity 94%) by the treatment with aluminum hydride in ether at room temperature.

A repetitive application of the above mentioned procedure resulted in the formation of the acetylenic ester $\underline{13}$ in 57% overall yield from the C_{12} -alcohol $\underline{12}$, and gave the vinyl sulfide $\underline{14}^4$) in 75% yield on treating of $\underline{13}$ with benzenethiol. The stereospecific methylation of $\underline{14}$ was effected by the reaction with the mixture

of methylmagnesium bromide and cuprous iodide in tetrahydrofuran at -78°C for 2 hr to give 87% yield of the desired trienic ester $\underline{15}$, synthetic precursor of C_{18} juvenile hormone, and the product so obtained was 93% pure by gas chromatographic analysis and exhibited fully consistent of n.m.r. and infrared spectra⁵⁾ with the structure 15.

The synthesis of the trienic ester $\underline{20}$, precursor of C_{17} juvenile hormone, was also achieved by the same reaction sequence from $\underline{10}$; (1) methylation to $\underline{16}$ (90% yield, purity 97%, bp 76-7°C/0.4mmHg), (2) reduction to $\underline{17}$ (99% yield), (3) propynylation followed by methoxycarbonylation to the ester $\underline{18}$ (45% yield from $\underline{17}$), (4) addition of benzenethiol (78% yield), and (5) methylation to the desired trienic ester 20^6) in 69% yield (purity 94%).

Further, methyl farnesoate was also synthesized in 38% overall yield (purity 95%) starting from geraniol.

The noteworthy feature of this sequence is the high stereospecificity and that wide variety of alkyl side chains could be introduced by the selective use of

various readily available Grignard reagents.

References

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- 1) S. Kobayashi and T. Mukaiyama, Chem. Lett., 705 (1974).
- 2) S. B. Bowlus and J. A. Katzenellenbogen, Tetrahedron Lett., 1277 (1973).
- 3) S. Kobayashi and T. Mukaiyama, Chem. Lett., 1097 (1973).
- 4) The configuration of β -phenylthio- α , β -ethylenic ester was determined by the chemical shift of the olefinic proton α to the methoxy-or ethoxycarbonyl group, since we have preliminary observed that α olefinic protons of Z-isomers absorb around δ 5.6 \sim 5.9, and those of E- isomers around δ 5.1 \sim 5.3, respectively.
- 5) n.m.r.; δ 0.97 (t, J = 7Hz, 6H), δ 1.66 (s, 3H), δ 1.8 \sim 2.3 (m, 12H), δ 2.15 (s, 3H), δ 3.62 (s, 3H), δ 5.03 (m, 2H), δ 5.60 (bs, 1H) i.r.: $\nu_{C=0}$ 1720 cm⁻¹, $\nu_{C=C}$ 1650 cm⁻¹.
- 6) n.m.r.; δ 0.97 (t, J = 7Hz, 3H), δ 1.60 (s, 3H), δ 1.64 (s 3H), δ 1.8 \sim 2.30 (m, 10H), δ 2.13 (s, 3H), δ 3.60 (s, 3H), δ 5.05 (m, 2H), δ 5.58 (bs, 1H) i.r.; $\nu_{C=0}$ 1720 cm⁻¹, $\nu_{C=C}$ 1650 cm⁻¹.

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